

Sultan Qaboos University
College of Science, Department of Chemistry

Chem3321/Chem3322
Introduction to Organic Chemistry/Organic Chemistry I

Spring 2009

Test 2

30 April, 2009

Test Duration: 90 minutes

Name: _____ ID: _____

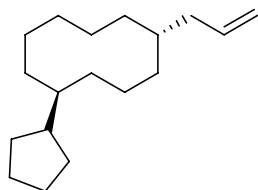
Question	Earned Mark	Maximum Mark
1		9
2		32
3		14
4		12
5		11
6		22
Total		100

Question 1

(9 marks)

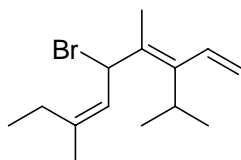
Name the following compounds according to IUPAC rules.

a.



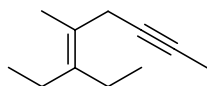
Name: _____

b.



Name: _____

c.



Name: _____

Question 2

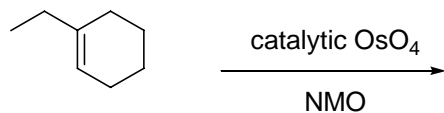
(32 marks)

Complete the following reactions by writing possible reactant(s), reagent(s) or product(s). If you think a certain reaction will not take place under the given conditions, write No Reaction.

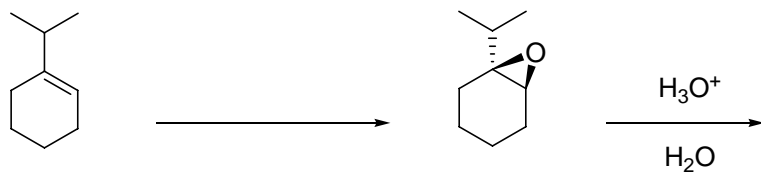
a.



b.



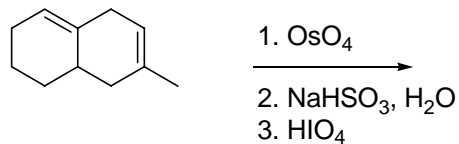
c.



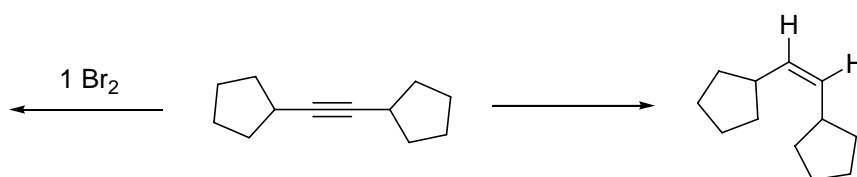
d.



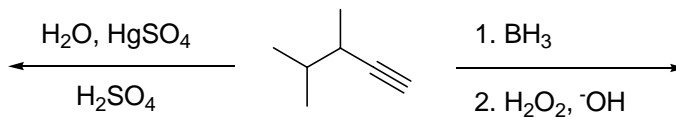
e.



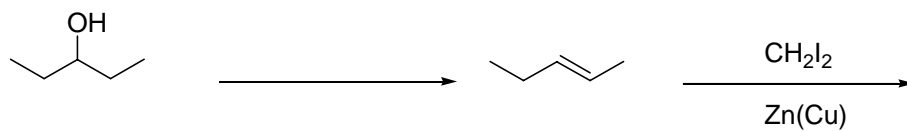
f.

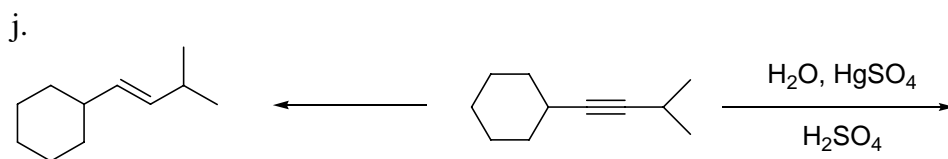
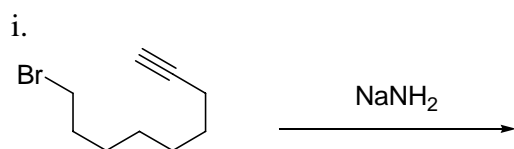


g.



h.





Question 3

(14 marks)

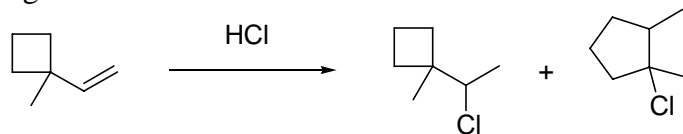
Consider trans-1-*t*-butyl-3-methylcyclohexane. Answer the following questions:

- Draw two chair conformations of this compound.
- Assuming that each diaxial $\text{CH}_3\text{-H}$ costs 3.8 kJ/mol and that each diaxial *t*-butyl-H costs 11.4 kJ/mol, calculate the cost of energy of each chair.
- Now, which of them is more stable and by how much? Briefly, explain the difference in stability.

Question 4

(12 marks)

Consider the following reaction:

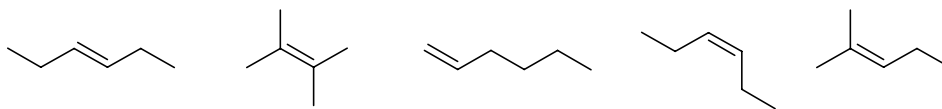


- In this reaction, what is the nucleophile and what is the electrophile?
- Propose a stepwise mechanism for the reaction using curved arrows.
- Two products are obtained in this reaction. Why? Provide an explanation.

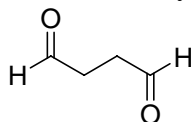
Question 5

(11 marks)

- a. Rank the following alkenes in order of increasing stability beginning with the least stable. Explain your answer briefly.



- b. A hydrocarbon has a formula of C_8H_{12} and reacts with 2 equivalents of H_2 in the presence of Pd/C as a catalyst. Upon ozonolysis with O_3 followed by Zn, H^+ treatment, the following dialdehyde is produced as the only product:



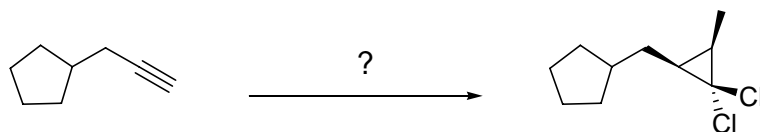
Propose a structure for the hydrocarbon.

Question 6

(22 marks)

Show how you would carry out the following transformations. More than one step is required in each case. Clearly outline all steps with reactants, reagents and intermediate products.

a.



b.

